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Note

Site-directed enzymatic α -(1 \rightarrow 3)-L-fucosylation of the tetrasaccharide Gal β (1 \rightarrow 4)GlcNAc β (1 \rightarrow 3)Gal β (1 \rightarrow 4)GlcNAc at the distal *N*-acetyllactosamine unit

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Glycans containing Lewis x, Gal $\beta(1 \to 4)$ [Fuc $\alpha(1 \to 3)$]GlcNAc $\beta 1 \to$, and related structures at their chain termini, are considered to be ligands for E-, L- and P-selectins [1-3]. Synthesis of Lewis x glycans, in sialylated or sulfated form, is of great interest, because of their potential as anti-inflammatory agents. The last step in the enzymatic synthesis of selectin ligand saccharides consists of α -(1 \to 3/4)-L-fucosylation, a reaction that can take place at all N-acetyllactosamine units of linear poly-N-acetyllactosamines [4], also in the tetrasaccharide Gal $\beta(1 \to 4)$ GlcNAc $\beta(1 \to 3)$ Gal $\beta(1 \to 4)$ GlcNAc (1) that reacts with human milk transferases (Niemelä et al., unpublished).

We have recently shown that human milk α - $(1 \rightarrow 3)$ -L-fucosyltransferase(s) do not react appreciably with N-acetyllactosamine residues that carry a β - $(1 \rightarrow 6)$ -linked GlcNAc branch at the galactose unit [5]. Here, this observation has been used to direct the enzymatic α - $(1 \rightarrow 3)$ -L-fucosylation of the divalent acceptor tetrasaccharide Gal β (1 \rightarrow 4)GlcNAc β (1 \rightarrow 3)Gal β (1 \rightarrow 4)GlcNAc (1) solely at the distal N-acetyllactosamine unit (see Fig. 1). The process involves (i) protection of the N-acetyllactosamine unit at

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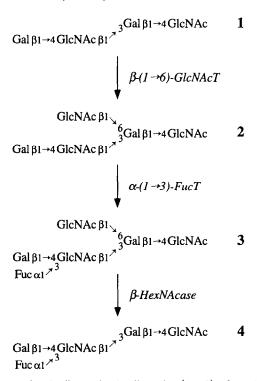


Fig. 1. The three reaction procedure leading to the site-directed α - $(1 \rightarrow 3)$ -L-fucosylation of glycan 1 to yield the Lewis x pentasaccharide 4. Abbreviations: β - $(1 \rightarrow 6)$ -GlcNAcT, β - $(1 \rightarrow 6)$ -D-N-acetylglucosaminyltransferase; α - $(1 \rightarrow 3)$ -FucT, α - $(1 \rightarrow 3)$ -L-fucosyltransferase; β -HexNAcase, β -D-N-acetylhexosaminidase.

the reducing end of 1 by mid-chain β - $(1 \rightarrow 6)$ -D-N-acetylglucosaminylation that generates $\operatorname{Gal}\beta(1 \rightarrow 4)\operatorname{GlcNAc}\beta(1 \rightarrow 3)[\operatorname{GlcNAc}\beta(1 \rightarrow 6)]\operatorname{Gal}\beta(1 \rightarrow 4)\operatorname{GlcNAc}$ (2), (ii) site-specific fucosylation of 2 by human milk α - $(1 \rightarrow 3)$ -L-fucosyltransferase(s) at the distal N-acetyllactosamine residue of 2, and (iii) removal of the protecting β - $(1 \rightarrow 6)$ -branched GlcNAc group with β -D-N-acetylhexosaminidase.

1. Results and discussion

Gal $\beta(1 \rightarrow 4)$ GlcNAc $\beta(1 \rightarrow 3)$ Gal $\beta(1 \rightarrow 4)$ GlcNAc (1) was converted to Gal $\beta(1 \rightarrow 4)$ GlcNAc $\beta(1 \rightarrow 3)$ [GlcNAc $\beta(1 \rightarrow 6)$]Gal $\beta(1 \rightarrow 4)$ GlcNAc (2) as described by Leppänen et al. [6], but by using the β -(1 \rightarrow 6)-D-N-acetylglucosaminyltransferase of rat serum [7] instead of human serum. The resulting pentasaccharide co-chromatographed on paper with authentic 2 [8]. Its ¹H-NMR spectrum (Fig. 2B and Table 1) revealed the H-1 resonance of the incoming β -(1 \rightarrow 6)-linked GlcNAc as one proton doublet at 4.585 ppm. The H-1 and H-4 resonances of the galactose residue 2 (for residue numbering, see Table 1) have both shifted to a higher field (compared to those of glycan

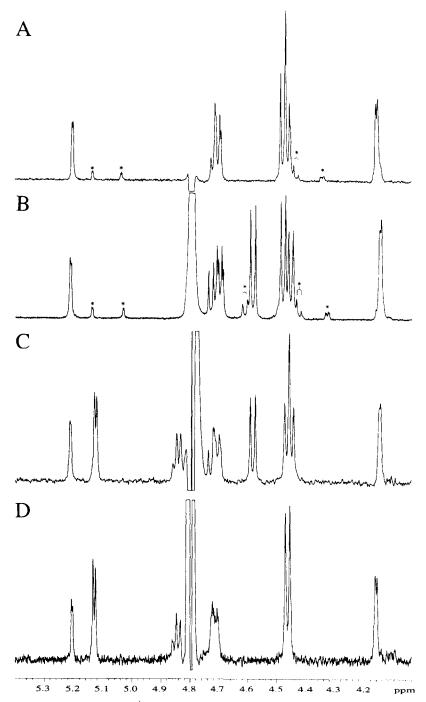


Fig. 2. Expansions of the 500 MHz ¹H-NMR spectra of (A) glycan 1, (B) glycan 2, (C) glycan 3 and (D) glycan 4. The signals marked by an asterisk arise from the reducing end epimers of glycans 1 and 2.

Reporter group	Residue a	Saccharides			
		1	2	3	4
H-1	1	5.205 (α)	5.212 (α)	5.211 (α)	5.204 (α)
		4.721 (β)	4.731 (β)	4.729 (B)	4.720 (β)
	2 b	4.465/4.462	4.454	4.450	4.462
	3 в	4.707/4.703	4.701/4.696	4.709/4.704	4.715/4.710
	4	4.480	4.481	4.465	4.462
	5	_	4.585	4.584	_
	6	_	_	5,125	5.128
H-4	2	4.159	4.149	4.148	4.159
H-5	6	-	-	4.840	4.840
Н-6	6	_	none	1.175	1.175

Table 1

H chemical shifts of structural reporter groups of saccharides 1-4 at 295 K

GleNAc
$$\beta$$
1 2 1
4 3 6Gal β 1 \rightarrow 4GleNAc β 1

Gal β 1 \rightarrow 4GleNAc β 1

Fuc α 1

1), indicating that this galactose is the site of substitution [9,10] and establishing the structure of the pentasaccharide as 2. The structure of 2 was further confirmed by ¹H and ¹³C 2D NMR experiments (Maaheimo et al., unpublished).

Glycan 2 was incubated with GDP-L-fucose and α -(1 \rightarrow 3)-L-fucosyltransferase(s) from human milk. The reaction mixture was separated by HPAE chromatography on CarboPac PA-1 column (Fig. 3A), revealing a major product (peak 1), which proved to represent hexasaccharide Gal $\beta(1 \rightarrow 4)$ [Fuc $\alpha(1 \rightarrow 3)$]GlcNAc $\beta(1 \rightarrow 3)$ [GlcNAc $\beta(1 \rightarrow 3)$]GlcNAc $\beta(1 \rightarrow 3)$ 6)]Gal $\beta(1 \rightarrow 4)$ GlcNAc (3); in addition, the reducing end epimer (peak 2), unreacted glycan 2 (peak 3) and its reducing end epimer (peak 4) were observed. The major peak migrated faster than glycan 2; this is characteristic to α -(1 \rightarrow 3)-fucosylated products [11]. The structure of 3 was established by H-NMR (Fig. 2C and Table 1). The spectrum reveals the H-1, H-5 and H-6 resonances of the incoming fucose at 5.125 ppm, 4.840 and 1.175 ppm (not shown), respectively. These chemical shifts are typical for α -(1 \rightarrow 3)-linked fucose [12]. The integrals of these signals indicated that there was only one fucose in glycan 3. The anomeric signals of GlcNAc residue 3 had shifted to somewhat lower field, while the H-1 signals of the reducing end GlcNAc were unaffected. Since also the anomeric signal of the galactose 4 experiences a notable upfield shift, the NMR data indicate that the fucose was bound to GlcNAc residue 3 rather than to residue 1. The structure of the glycan 3 was also supported by the MALDI-TOF mass spectrum (Fig. 4): a major peak assigned to $(M + Na)^+$ was

^a Numbering of the residues is as follows:

^b The two chemical shift values given arise from signals representing the α - and β -pyranosic forms of the oligosaccharide.

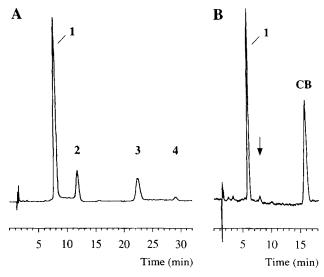


Fig. 3. High-pH anion exchange chromatograms. (A) The desalted and gel filtrated purified products of a α -(1 \rightarrow 3)-L-fucosyltransferase reaction of glycan 2. Peak 1 represents the fucosylated glycan 3, while peak 3 is the unreacted glycan 2. Peaks 2 and 4 are the reducing end epimers of glycans 3 and 2, respectively. Elution system: 10 min isocratic with 100 mM NaOH, then linear gradient of increasing concentration of NaOAc (rate of increase 1 mM/min). (B) A small aliquot from β -D-N-acetylhexosaminidase digest of glycan 3 after gel filtration. Peak 1 is the Lewis x saccharide 4 and CB internal marker, cellobiose. Arrow shows the elution position of glycan 3. Elution with 100 mM NaOH. No reducing end epimer of glycan 4 was formed, probably because the enzymatic cleavage reaction was performed at pH 4.0.

observed at m/z 1120.7 (calculated m/z 1121.0), while a minor peak assigned to $(M + K)^+$ was at m/z 1136.8 (calculated m/z 1137.0).

Hydrolysis of glycan 3 with β -D-N-acetylhexosaminidase converted it into Gal $\beta(1 \rightarrow 4)$ [Fuc $\alpha(1 \rightarrow 3)$]GlcNAc $\beta(1 \rightarrow 3)$ Gal $\beta(1 \rightarrow 4)$ GlcNAc (4). This is shown by a run of small aliquot of the reaction product(s) in HPAE chromatography (Fig. 3B), where peak 1 is the glycan 4, while the arrow is showing the position of glycan 3. When

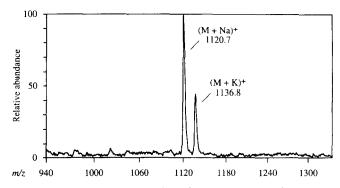


Fig. 4. MALDI-TOF mass spectrum of glycan 3 in the mass range of m/z 940–1360.

compared to the NMR spectrum of 3, the most striking feature in the spectrum of 4 (Fig. 2D and Table 1) is the complete absence of doublet at 4.584 ppm, which is the H-1 resonance of the β -(1 \rightarrow 6)-linked GlcNAc (residue 5). The H-1 signals of the β -(1 \rightarrow 3)-linked GlcNAc (residue 3) have shifted to lower field and are now overlapping with those of the β -anomer of the reducing end GlcNAc; the two galactose H-1s resonate now at identical frequencies at 4.462 ppm. Comparison of the spectrum of 4 with that of 1 reveals that the anomeric signal of GlcNAc residue 3 has shifted to a lower field, while the H-1 signal of GlcNAc residue 1 is unshifted, further establishing that the fucose is linked to GlcNAc residue 3. MALDI-TOF mass spectrum of glycan 4 revealed that one N-acetylhexosamine residue had been hydrolyzed from the glycan 3; a peak assigned to $(M + Na)^+$ (calculated m/z 917.8) was observed at m/z 917.6. The structure of glycan 4 was further established by β -D-galactosidase digestion, which left the glycan intact (data not shown), as expected, because β -D-galactosidase does not hydrolyze Lewis x structures [13].

The present three-reaction-procedure bears close resemblance to a similar process where the α -(1 \rightarrow 3)-L-fucosylation is directed away from the distal *N*-acetyllactosamine unit by temporary α -(2 \rightarrow 6)-sialylation [14]. The sialyl group at 6'-position of the 'outer' *N*-acetyllactosamine unit prevents the fucosylation at this *N*-acetyllactosamine [5,15].

The success of the present protection-reaction-deprotection procedure depends on the specific mode of action of two enzymes: (i) β -(1 \rightarrow 6)-D-N-acetylglucosaminyltransferase of rat serum, that does not act on the distal N-acetyllactosamine units but rather on the inner chain units [7]; (ii) α -(1 \rightarrow 3)-L-fucosyltransferase that does not act on branching N-acetyllactosamine units but acts on linear ones [5]. The value of this process lies in the site-specific formation of a pure Lewis x structure containing glycan 4, carrying the α -(1 \rightarrow 3)-L-fucosylated N-acetyllactosamine unit solely at the distal position. The process can also be used to synthesize pure sialyl Lewis x (sLex) glycans; we have synthesized branched structures like NeuAc α (2 \rightarrow 3)Gal β (1 \rightarrow 4)GlcNAc β (1 \rightarrow 3)[NeuAc α (2 \rightarrow 3)Gal β (1 \rightarrow 4)GlcNAc β (1 \rightarrow 4)GlcNAc, which are exclusively fucosylated at the distal N-acetyllactosamine units, rather than at the branching N-acetyllactosamine [16]. Hence, pure Lewis x structures and their sialyl derivatives, free from VIM-2 type isomers or difucosylated products, can be obtained by the present process.

Other methods to obtain the same goal appear to be emerging: (i) Howard et al. [17] have reported that nLc_6 , $Gal\beta(1\rightarrow 4)GlcNAc\beta(1\rightarrow 3)Gal\beta(1\rightarrow 4)GlcNAc\beta(1\rightarrow 3)Gal\beta(1\rightarrow 4)Glc\beta(1\rightarrow 1)Cer$, is fucosylated mainly at the external N-acetyllactosamine unit by the α -(1 \rightarrow 3)-L-fucosyltransferase II of CHO-LEC12 cells. (ii) With glycolipids as acceptors, the proper choice of detergent can also lead to external α -(1 \rightarrow 3)-L-fucosylation [18], but it is not clear whether sterically unhindered, non-micellar acceptors would react similarly. (iii) A chromatographic method capable of separating glycan 4 from its isomer $Gal\beta(1\rightarrow 4)GlcNAc\beta(1\rightarrow 3)Gal\beta(1\rightarrow 4)[Fuc\alpha(1\rightarrow 3)]GlcNAc$ has been developed in our laboratory (Niemelä et al., unpublished). Hence, glycan 4 can be directly isolated from the monofucosylated products generated by partial α -(1 \rightarrow 3)-L-fucosylation of glycan 1 with human milk α -(1 \rightarrow 3)-L-fucosyltransferase(s).

2. Experimental

Oligosaccharide primer.—The synthesis of Gal $\beta(1 \rightarrow 4)$ GlcNAc $\beta(1 \rightarrow 3)$ Gal $\beta(1 \rightarrow 4)$ GlcNAc (1) was performed essentially as described in [19], using four-fold molar excess of UDP-D-galactose (Sigma, MO, USA).

Enzymatic methods.— α - $(1 \rightarrow 3)$ -L-Fucosyltransferase(s) (human milk) reactions were performed as described in [5,20]. β - $(1 \rightarrow 6)$ -D-N-acetylglucosaminyltransferase reactions were performed in the presence of 20 mM EDTA, 200 mM D-galactose and 60 mM D-galactonic acid γ -lactone, essentially as described in ref. [6], with the exception that rat serum was used as enzyme source. Hydrolysis with β -D-galactosidase (Jack beans, EC 3.2.1.23; Sigma) [21] and β -D-N-acetylhexosaminidase (Jack beans, EC 3.2.1.23; Sigma) [6] was performed as described.

Chromatographic methods.—Gel filtration was performed in a column of Superdex 75 HR 10/30 (Pharmacia, Sweden), with water as the eluant at a flow rate of 1 mL/min. The eluant was monitored at 205 nm, and oligosaccharides were quantified against external 2-acetamido-2-deoxy-p-glucose (p-GlcNAc). High-pH anion exchange chromatography with pulsed amperometric detection (HPAEC-PAD) was carried out with a Dionex series 4500i HPLC system (Dionex, CA, USA) with a CarboPac PA-1 column (4 × 250 mm), as described [22]. The column was equilibrated with 100 mM NaOH and run as follows: either isocratic with 100 mM NaOH or first 10 min isocratic with 100 mM NaOH followed with linear gradient of NaOAc ending to 100 mM NaOAc-100 mM NaOH at 110 min. Peaks were collected manually, neutralized with 0.4 M acetic acid and dried in a vacuum lyophilizer. The oligosaccharides were desalted by filtration in water through AG-1 (AcO⁻) and AG-50W (H⁺) (Bio-Rad, CA, USA).

 $^{\prime}$ H-NMR spectroscopy.—The samples were twice lyophilized from D₂O, 99.96% D (Cambridge Isotope Laboratories, MA, USA), and finally dissolved in 600 μL of D₂O, 99.996% D (CIL). The spectra were recorded on Varian Unity 500 spectrometer at 295 K using a modification of the WEFT sequence [23] for suppression of the residual HDO signal. The chemical shifts were referenced to internal acetone, 2.225 ppm.

MALDI-TOF MS.—Matrix-assisted laser desorption/ionization time of flight mass spectrometry was performed in the positive ion mode with irradiation from a nitrogen laser (337 nm) and 2,5-dihydroxybenzoic acid as the matrix with the Vestec VT-2000 linear time-of-flight instrument operated at 30 kV accelerating voltage. External calibration was used; this method has an accuracy of $\pm 0.1\%$ (± 2 u at m/z 2000).

Acknowledgements

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References

- [1] T. Feizi, Curr. Opin. Struct. Biol., 3 (1993) 701-710.
- [2] S.D. Rosen and C.R. Bertozzi, Curr. Opin. Cell Biol., 6 (1994) 663-673.
- [3] P.J. Green, C.-T. Yuen, R.A. Childs, W. Chai, M. Miyasaka, R. Lemoine, A. Lubineau, B. Smith, H. Ueno, K.C. Nicolaou, and T. Feizi, Glycobiology, 5 (1995) 29-38.
- [4] T. de Vries and D.H. Van den Eijnden, Histochem. J., 24 (1992) 761-770.
- [5] R. Niemelä, J. Natunen, E. Brotherus, A. Saarikangas, and O. Renkonen, Glycoconjugate J., 12 (1995) 36-44.
- [6] A. Leppänen, L. Penttilä, R. Niemelä, J. Helin, A. Seppo, S. Lusa, and O. Renkonen, Biochemistry, 30 (1991) 9287–9296.
- [7] J. Gu, A. Nishikawa, S. Fujii, S. Gasa, and N. Taniguchi, J. Biol. Chem., 267 (1992) 2994-2999.
- [8] O. Renkonen, A. Leppänen, R. Niemelä, A. Vilkman, J. Helin, L. Penttilä, H. Maaheimo, A. Seppo, and J. Suopanki, Biochem. Cell Biol., 70 (1992) 86-89.
- [9] D.M. Whitfield, H. Pang, J.P. Carver, and J.J. Krepinsky, Can. J. Chem., 68 (1990) 942-952.
- [10] H. Maaheimo, L. Penttilä, and O. Renkonen, FEBS Lett., 349 (1994) 55-59.
- [11] M.R. Hardy and R.R. Townsend, Proc. Natl. Acad. Sci. U.S.A., 85 (1988) 3289-3293.
- [12] J.P. Kamerling and J.F.G. Vliegenthart, in L.J. Berliner and J. Reuben (Eds.), Biol. Magn. Res., Vol. 10, Plenum, New York, 1992, pp 1–287.
- [13] M. Arakawa, S.-I. Ogata, T. Muramatsu, and A. Kobata, J. Biochem., 75 (1974) 707-714.
- [14] M.A. Kashem, C. Jiang, A.P. Venot, and G.R. Alton, Carbohydr. Res., 230 (1992) C7-C10.
- [15] J.C. Paulson, J.-P. Prieels, L.R. Glasgow, and R.L. Hill, J. Biol. Chem., 253 (1978) 5617-5624.
- [16] J.P. Turunen, M.-L. Majuri, S. Tiisala, T. Paavonen, M. Miyasaka, K. Lemström, A. Seppo, L. Penttilä, O. Renkonen, and R. Renkonen, J. Exp. Med., 182 (1995) 1133–1142.
- [17] D.R. Howard, M. Fukuda, M.N. Fukuda, and P. Stanley, J. Biol. Chem., 262 (1987) 16830-16837.
- [18] E.H. Holmes and B.A. Macher, Arch. Biochem. Biophys., 301 (1993) 190-199.
- [19] O. Renkonen, L. Penttilä, R. Niemelä, and A. Leppänen, Glycoconjugate J., 8 (1991) 376-380.
- [20] M.M. Palcic, A.P. Venot, R.M. Ratcliffe, and O. Hindsgaul, Carbohydr. Res., 190 (1989) 1-11.
- [21] O. Renkonen, J. Helin, L. Penttilä, H. Maaheimo, R. Niemelä, A. Leppänen, A. Seppo, and K. Hård, Glycoconjugate J., 8 (1991) 361–367.
- [22] J. Helin, H. Maaheimo, A. Seppo, A. Keane, and O. Renkonen, Carbohydr. Res., 266 (1995) 191-209.
- [23] K. Hård, G. van Zadelhoff, P. Moonen, J.P. Kamerling, and J.F.G. Vliegenthart, Eur. J. Biochem., 209 (1992) 895-915.